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Cordiachromes: A New Group of Terpenoid Quinones from Cordia spp.

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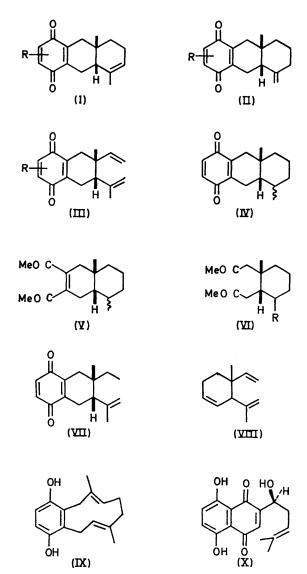
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Summary Cordia millenii contains six optically inactive benzoquinones, cordiachromes (A)—(F); (C) and (F) have a benzogeijerene skeleton and the others have a tricyclic benzocogeijerene structure; they are probably derived from a benzenoid precursor and geranyl pyrophosphate.

WOODWORKERS handling *Cordia* spp. (Boraginaceae) are prone to contract skin allergies.¹ This led us to examine the constituents of these timbers and a cold chloroform extract of the heartwood of *C. millenii* has been found to contain at least six quinones, cordiachromes (A—F), separable by extensive chromatography on silica gel-silver nitrate plates.

Cordiachromes (A—C), $C_{16}H_{18}O_2$, are isomeric. They give positive Craven tests, form leuco-diacetates, and show λ_{max} (EtOH) 250 and 348 nm (log ϵ 4·1 and 2·9), λ_{max} (EtOH-BH₄⁻⁾ 292 nm, ν_{co} ca. 1650 cm⁻¹, and τ (CDCl₃) ca. 3.30 (2H, s). Thus they are 2,3-dialkylbenzoquinones, the non-quinonoid moieties being C10H16. Each quinone contains a tertiary methyl group (τ ca. 9.0) but they differ in their olefinic absorption; in (A) (a yellow oil) a broad singlet (3H) at τ 8.27 is coupled to a vinylic multiplet (1H) at 4.69, while (B) (golden needles, m.p. 64-65°) shows two singlets (each 1H) at τ 5.27 and 5.62. Further, (B) absorbs strongly at 913 cm⁻¹, gives formaldehyde on ozonolysis, and on fusion with zinc dust yields an aromatic hydrocarbon $[\lambda_{max} (C_6 H_{12}) 238, 260, 332, 350, 367, and 386 nm], probably$ a dimethylanthracene. As both (A) and (B) gave the same dihydro-derivative after hydrogenation and reoxidation with silver oxide, we formulate them as (I; R = H) and (II; R = H), respectively, (ignoring stereochemistry). The dihydro-derivative (IV) is clearly a pair of epimers, the overlapping methyl signals at τ 9.14 (6H) (CDCl₃) separating into two sets of overlapping signals at τ 9.32 and 9.40 in C_6D_6 . Oxidation of (IV) with alkaline hydrogen peroxide



afforded, after methylation, the maleic ester (V), $C_{16}H_{24}O_4$, consistent with all spectroscopic parameters, while ozonolysis of (IV) followed by acidic hydrogen peroxide, and then methylation, gave (VI; R = Me), $C_{14}H_{20}O_4$. The mass spectrum of (VI; R = Me) is virtually identical with that of (VI; R = H) displaced by 14 m.u. The *cis*-geometry of (VI; R = Me), and hence of (I) and (II), follows from the chemical shift of the tertiary methyl at τ 8.95 identical with that in the spectrum of cis-(VI; R = H)² whereas for the trans-isomer³ of (VI; R = H) the signal appears at τ 9.07.

Cordiachrome C (III), a yellow oil, contains five olefinic protons (besides QH) and shows, inter alia, y 913 and 895 cm⁻¹, τ (CDCl₃) 4.08 (1H, dd, J 11 and 17 Hz, $-CH = CH_2$, 4.98 (1H, d, J 11 Hz, $-CH = CH_2$), 5.10 (1H, d, $J 17 \text{ Hz}, -CH = CH_2$, 5.08 and 5.22 (each 1H, s, $> = CH_2$), and 8.23 (3H, s, $CH_3-C=C$). The tertiary methyl protons resonate at τ 8.89 which suggests that (C) is also cis-(III) but this point is not established. Hydrogenation in acetic acid (followed by reoxidation) gave a tetrahydro-derivative, but in ethyl acetate the dihydro-derivative (VII) was obtained [v 895 cm⁻¹, τ 5.12 and 5.30 (each 1H)].

Cordiachromes (D--F), C17H20O3, are methoxylated analogues of (A-C), *i.e.* (I, II, and III; R = OMe). Corresponding pairs are spectroscopically identical except that (D—F) show methoxy-signals at τ 6.20 and only one quinonoid proton at $\tau 4.12$.

Like geijerene (VIII),⁴ the cordiachromes are racemic and probably arise from an optically inactive⁵ "benzogermacrane"-type precursor, e.g. (IX),6 derived from geranyl pyrophosphate and an aromatic unit. It is interesting that the only other quinones known in the Boraginaceae are naphthazarins (X and close relatives),7 and it has been established recently⁸ that alkannin (X) is biosynthesised from p-hydroxybenzoic acid and mevalonate. It appears that both the cordiachromes and alkannin are formed by attachment of a geranyl side chain to an aromatic precursor, followed by oxidation of an allylic methyl group which then cyclises.

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